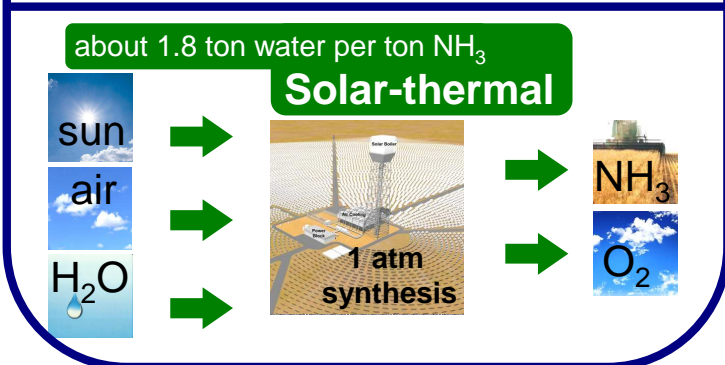
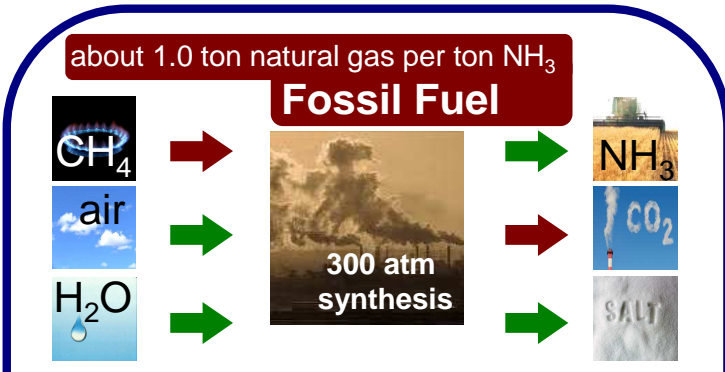
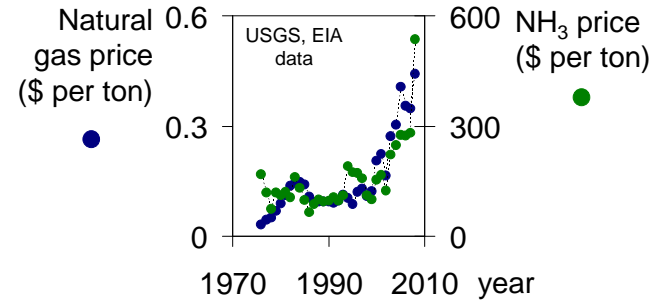


Motivation and impact

- About 50% of the global food supply depends on ammonia (NH₃) fertilizer that is synthesized industrially (about 130 million tons/year) with 5% of the global natural gas production.
- This work proposes ammonia production from sunlight, water and air and analyzes its technical (experimental reaction yield and kinetics studies) and economic feasibility (mass and energy balance based process modeling and net present value analysis).



Experimental

Economic analysis

Total plant costs (ideal case, in % of 0.55 million \$)

Conclusions and Outlook

- The process is technical feasible with a molybdenum reactant. To increase the 13 mol% yield of NH₃ and the 25 mol% nitride yield further work optimizing the reactant composition is needed.
- Under conservative assumptions the process may sell NH₃ for 534 \$ per ton (i.e., near the current market price), producing at a capacity between 160 to 900 tons NH₃ per day at optimum.

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